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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO
10/712,326	11/12/2003	Rao Annapragada	LAM-P-1031	2008
48008 7590 05/10/2007 VIRTUAL LEGAL, P.C. MICHAEL A. KERR			EXAMINER	
			NGUYEN, THANH T	
3476 EXECUTIVE POINTE WAY, UNIT 16 CARSON CITY, NV 89706		NIT 16	ART UNIT	PAPER NUMBER
O/MOON OIT			2813	
			MAIL DATE	DELIVERY MODE
			05/10/2007	PAPER

Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

		Application No.	Applicant(s)			
Office Action Summary		10/712,326	ANNAPRAGADA ET AL.			
		Examiner	Art Unit			
		Thanh T. Nguyen	2813			
Period fo	The MAILING DATE of this communication a	appears on the cover sheet with	the correspondence address			
A SH THE - Exte after - If the - If NC - Failu Any	ORTENED STATUTORY PERIOD FOR REI MAILING DATE OF THIS COMMUNICATION insions of time may be available under the provisions of 37 CFR SIX (6) MONTHS from the mailing date of this communication. It period for reply specified above is less than thirty (30) days, a properiod for reply is specified above, the maximum statutory period for reply within the set or extended period for reply will, by state reply received by the Office later than three months after the material patent term adjustment. See 37 CFR 1.704(b).	N. 1.136(a). In no event, however, may a reply reply within the statutory minimum of thirty (3 dod will apply and will expire SIX (6) MONTH: tute, cause the application to become ABAN	be timely filed 0) days will be considered timely. S from the mailing date of this communication. DONED (35 U.S.C. § 133).			
Status						
1)[Responsive to communication(s) filed on 13	3 February 2007.				
		his action is non-final.				
3)	Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under <i>Ex parte Quayle</i> , 1935 C.D. 11, 453 O.G. 213.					
Disposit	ion of Claims					
5)□ 6)⊠ 7)□ 8)□	Claim(s) 1-21 is/are pending in the application 4a) Of the above claim(s) is/are with the claim(s) is/are allowed. Claim(s) 1-21 is/are rejected. Claim(s) is/are objected to. Claim(s) are subject to restriction and the companion papers	Irawn from consideration.				
9) The specification is objected to by the Examiner.						
10) ☐ The drawing(s) filed on is/are: a) ☐ accepted or b) ☐ objected to by the Examiner.						
Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).						
Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d). 11) The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.						
Priority ι	ınder 35 U.S.C. § 119					
a)(Acknowledgment is made of a claim for foreignal All b) Some * c) None of: 1. Certified copies of the priority docume 2. Certified copies of the priority docume 3. Copies of the certified copies of the papplication from the International Bure See the attached detailed Office action for a light service.	ents have been received. ents have been received in App riority documents have been re eau (PCT Rule 17.2(a)).	lication No ceived in this National Stage			
Attachmen	t(s) e of References Cited (PTO-892)		mary (PTO-413)			
3) 🔲 Infon	e of Draftsperson's Patent Drawing Review (PTO-948) mation Disclosure Statement(s) (PTO-1449 or PTO/SB/ r No(s)/Mail Date		lail Date mal Patent Application (PTO-152)			

DETAILED ACTION

Response to Arguments

Applicant's arguments filed 2/13/07 have been fully considered but they are not persuasive.

Claim Rejections - 35 USC § 103

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

Claims 1-21 are stand rejected under 35 U.S.C. 102(e) as being anticipated by Chooi et al. (U.S. Patent No. 6,465,888) in view of Morrow et al. (U.S. Patent Publication No. 2002/0081854).

Referring to figures 2a-4f, Chooi et al. teaches a method of removing a photoresist layer (see col. 7, lines 59-60) form an integrated circuit (IC) structure with little or no etching of an exposed barrier layer comprising an integrated circuit (IC) structure having an etched dielectric layer with an exposed barrier layer, wherein the dielectric layer comprises silicon and oxygen (230, see col. 7, lines 47-54) and the barrier layer comprises silicon nitride or silicon carbide (215, see col. 7, lines 20-33), the method comprising:

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Firstly, feeding a first gas mixture into a reactor wherein the first gas mixture comprises carbon monoxide (CO)(see col. 8, lines 17-33), after etching the dielectric layer (230) and exposing the barrier layer (215, see figure 2b);

Secondly, energizing the oxidizing gas mixture having carbon monoxide (CO) to generating a plasma in the reactor (see col. 8, lines 17-33, noted that gas have to flow in the chamber and plasmanizing); and

Selectively removing the photoresist layer with little or no etching of the exposed barrier layer (see figure 2b, col. 8, lines 11-16), thereby minimizing the loss of the exposed barrier material during removing the photoresist layer. Noted that since removing the photoresist by ashing without removing anything inside of the opening would minimize the loss of the barrier material.

Regarding to claim 2, dielectric material is silicon dioxide (230, see col. 7, lines 47-54).

Regarding to claim 3, the first gas mixture further comprises oxygen (O₂) (see col. 8, lines 17-33).

Regarding to claim 4, the first gas mixture further comprises nitrogen (N_2) (see col. 8, lines 17-33).

Regarding to claims 5, 11, 15, the first gas mixture further comprise the gas mixture selected from the group consisting of oxygen, nitrogen, nitrogen/oxygen, nitrous oxide, ammonia, nitrogen/hydrogen, and water vapor (see col. 8, lines 17-33).

Regarding to claims 6, 12, 17, etched dielectric material is composed of a material selected from the group consisting of silicon dioxide, silicon oxide, organosilicate glass, and fluorinate silicate glass (see col. 7, lines 34-54).

Regarding to claims 7, 13, 18, cap layer located between the dielectric and the photoresist, the cap layer is composed of a material selected from the group consisting of silicon dioxide, silicon oxynitride, silicon carbide and silicon nitride (235, silicon nitride, see col. 7, lines 54-58).

Regarding to claims 8, 14, reactor used to remove the photoresist from the IC structure is also used to etch the dielectric (see col. 8, lines 1-16).

Regarding to claim 9, 21, a third layer that includes a conductive interconnect (210) that abuts the barrier layer (215) and the second dielectric material (220) adjacent the conductive interconnect, the barrier (215) between the etched first dielectric layer (230) and the third layer (210).

Regarding to claims 10, 16, 19, 21, the first dielectric layer (230) and the second dielectric layer (220) is comprised of materials that include silicon and oxygen (see col. 7, lines 34-54, noted that silicon oxide includes silicon and oxygen).

Chooi et al. teaches etching the photoresist film by using carbon monoxide gas (CO) (see col. 8, lines 17-33). However, the reference does not teach removing the photoresist film from the surface of the structure by using carbon monoxide gas.

Morrow et al. teaches removing the photoresist film by using mixture of carbon monoxide gas (CO), oxygen and nitrogen gas from the surface of the structure (see figure 5e, paragraph# 54). Noted that the same gas would inherently provide the same function as minimizing the loss of the exposed barrier during the removal of the photoresist film.

Therefore, it would have been obvious to a person of ordinary skill in the requisite art at the time of the invention was made would removing the photoresist film by using carbon

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monoxide gas (CO) from the surface of the structure in process of Chooi et al. as taught by

Morrow et al. because removing the photoresist film by using mixture of carbon monoxide gas

(CO), oxygen and nitrogen gas from the surface of the structure would prevent attack or damage
to the bottom layer or the side wall of the structure.

Response to Arguments

Applicant's arguments filed 2/13/07 have been fully considered but they are not persuasive.

Applicant contends that Chooi et al. does not teach the use of an oxidizing gas mixture comprising carbon monoxide (CO) to remove the photoresist when the dielectric has been previously etched to expose the barrier layer. In response to applicant that Chooi et al. clearly teaches etching the dielectric layer (230) and exposing the barrier layer (215, see figure 2b) then feeding the oxidizing gas mixture of CO (see col. 8, lines 17-33) then removing the photoresist film with oxygen gas (see figure 2b, col. 8, lines 11-16). Chooi et al. also teach a method of etching the photoresist film by using CO (see col. 8, lines 17-33). However, Chooi et al. does not specifically teach removing the photoresist film by using CO. Examiner relies on Morrow to show the only feature that it is known in the semiconductor art to remove the photoresist film by using CO (see figure 5e, para# 54). Therefore, Chooi in view of Morrow do teach oxidizing gas mixture comprising carbon monoxide (CO) to remove the photoresist when the dielectric has been previously etched to expose the barrier layer.

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Applicant also contends that Chooi et al. would etch the barrier during the removal of the photoresist layer by ashing because oxidizing gas mixture reacts with fluorinated polymer would etch the barrier layer. In response to applicant that Chooi et al. in col. 8, lines 11-16, teaches that removal of the photoresit layer includes one or more of the following: fluorocarbon, such as CF4, C4F8, hydrocarbon, fluorine-substituted hydrocarbons, fluorosulfur, chlorine, hydrogen bromide, oxygen, nitrogen, argon and carbon monoxide (see Chooi, col. 8, lines 3-7). This can be means that removal of the photoresist film can be used any single or combination of gases such as chlorine, hydrogen bromide, oxygen, nitrogen, argon and carbon monoxide wherein fluorocarbon, such as CF4, C4F8, hydrocarbon, fluorine-substituted hydrocarbons, fluorosulfur gas does not have to include during the removal of the photoresit film. Therefore, it would not remove the barrier film.

Conclusion

THIS ACTION IS MADE FINAL. Applicant is reminded of the extension of time policy as set forth in 37 CFR 1.136(a).

A shortened statutory period for reply to this final action is set to expire THREE MONTHS from the mailing date of this action. In the event a first reply is filed within TWO MONTHS of the mailing date of this final action and the advisory action is not mailed until after the end of the THREE-MONTH shortened statutory period, then the shortened statutory period will expire on the date the advisory action is mailed, and any extension fee pursuant to 37 CFR 1.136(a) will be calculated from the mailing date of the advisory action. In no event,

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however, will the statutory period for reply expire later than SIX MONTHS from the mailing date of this final action.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Thanh Nguyen whose telephone number is (571) 272-1695, or by Email via address Thanh.Nguyen@uspto.gov. The examiner can normally be reached on Monday-Thursday from 6:00AM to 3:30PM.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Carl Whitehead, Jr., can be reached on (571) 272-1702. The fax phone number for this Group is (571) 273-8300.

Any inquiry of a general nature or relating to the status of this application or proceeding should be directed to the Group receptionist whose telephone number is (703) 308-0956 (See MPEP 203.08).

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pairdirect.uspto.gov. Should you have questions on access to thy Private PAIR system, contact the Electronic Business center (EBC) at 866-217-9197 (toll-free).

Thanh Nguyen
Patent Examiner

Patent Examining Group 2800